

# Superconductivity of monolayer Mo<sub>2</sub>C: The key role of functional groups

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Monolayer Mo<sub>2</sub>C is a new member of two-dimensional materials. Here the electronic structure and lattice dynamics of monolayer Mo<sub>2</sub>C are calculated. According to the electron-phonon interaction, it is predicted that monolayer Mo<sub>2</sub>C could be a quasi-two-dimensional superconductor and the effects of functional-groups are crucially important considering its unsaturated surface. Despite the suppressed superconductivity by chalcogen adsorption, our most interesting prediction is that the electron-phonon interaction of monolayer Mo<sub>2</sub>C can be greatly enhanced by bromine absorption, suggesting that Mo<sub>2</sub>CBBr<sub>2</sub> as a good candidate for a nanoscale superconductor. *Published by AIP Publishing*. [<http://dx.doi.org/10.1063/1.4974085>]

## I. INTRODUCTION

Two-dimensional (2D) graphene-like carbides and carbonitrides (MXenes, e.g., Ti<sub>2</sub>C and Nb<sub>2</sub>C) have attracted enormous interest for their novel chemical and physical properties, since they were successfully produced by etching the A layers of MAX phase (M is an early transition metal, A is an element from group IIIA or IVA, and X is carbon or nitrogen).<sup>1-3</sup> Due to the unsaturated surface with unpaired electrons, the surfaces of MXenes always easily adsorb various functional groups (e.g., F, O, or/and OH group) during etching, thus the chemical and physical properties are varying with various adsorptions. For this reason, MXenes and their functionalized ones have been widely investigated regarding the magnetism, electronic structures, as well as catalytic properties and energy storage.<sup>4-8</sup>

Recently, 2D layered Mo<sub>2</sub>C as a new member of MXenes was formed from Mo<sub>2</sub>Ga<sub>2</sub>C thin films. The metallic nature of the Mo–Ga bond is weaker than the Mo–C bond which has a mixed covalent/metallic/ionic character, thus monolayer Mo<sub>2</sub>C can be produced by selectively etching the Ga layer.<sup>9</sup> Structurally, monolayer Mo<sub>2</sub>C is constructed by the Mo–C–Mo sandwich, as shown in Fig. 1(a), which looks similar to 1T-MoS<sub>2</sub>. Khazaei *et al.* studied the thermoelectric properties of monolayer Mo<sub>2</sub>C, which was found to be a promising candidate as a high-performance thermoelectric material.<sup>10</sup> In addition, the structural, electrical, thermal, and mechanical properties of monolayer Mo<sub>2</sub>C were also studied.<sup>11</sup>

Superconductivity in ultrathin films owns promising future for applications, e.g., superconducting computational devices, thus great efforts have been made to discover 2D superconductors.<sup>12-14</sup> Considering that monolayer Mo<sub>2</sub>C is non-magnetic and appears strong metallicity (according to the first-principles study),<sup>10</sup> it brings the opportunity to be an ultrathin superconductor. However, although the superconductivity was discovered in bulk  $\alpha$ -Mo<sub>2</sub>C (an allotrope of 2D layered Mo<sub>2</sub>C), its superconducting transition temperature ( $T_C$ ) was depressed from 4 K (thick film) to near 0 K when its

thickness is less than 3.5 nm.<sup>15,16</sup> Moreover, analogous to other members of MXenes, the surfaces of monolayer Mo<sub>2</sub>C are easily covered by functional groups, and Khazaei *et al.* indicated that the full adsorption of functional groups is more stable than the cases of partial adsorption.<sup>17</sup> For multilayer Mo<sub>2</sub>C MXenes, they (stacking like MoS<sub>2</sub>) seem to be impossible due to its unsaturated surfaces. In experiments, the possible stackings are –Mo<sub>2</sub>CX<sub>2</sub>–Mo<sub>2</sub>CX<sub>2</sub>– and Mo–C–Mo–C–Mo. For the former, the effects of functional groups are unavoidable. For the latter, the stoichiometry has changed. These facts motivate us to investigate the superconductivity of the pure Mo<sub>2</sub>C monolayer and those with various functional groups.

In this work, the lattice dynamics and electron-phonon coupling (EPC) of monolayer Mo<sub>2</sub>C and functionalized ones have been studied via first-principles density functional theory (DFT) and density function perturbation theory (DFPT). Our calculations find that the strong EPC in monolayer Mo<sub>2</sub>C may lead to superconductivity, while the oxidation would make its superconductivity disappear. Besides, the superconducting  $T_C$  is also obviously suppressed by the absorption of sulfur or selenium. In contrast, the EPC of monolayer Mo<sub>2</sub>C can be greatly enhanced by bromine absorption, leading to a predicted  $T_C$  up to 12 K.

## II. MODEL AND METHODS

The electronic structure calculations have been performed using the Vienna *ab initio* simulation package (VASP) with a projector augmented wave method.<sup>18,19</sup> Generalized gradient approximation of Perdew-Burke-Ernzerhof (GGA-PBE) is used with a cutoff energy of 600 eV. The spin-orbit coupling (SOC) is also included in electronic structure calculations. The phonon dispersion calculations are carried out using the ultrasoft pseudo-potential (including the semicore electrons as valence electrons in the case of Mo) as implemented in the PWSCF program of the Quantum-ESPRESSO distribution, which are calculated within the framework of DFPT.<sup>20</sup> In the DFPT calculation, the GGA-PBE formulation is also used with a cutoff energy of 35 Ry for the expansion of the

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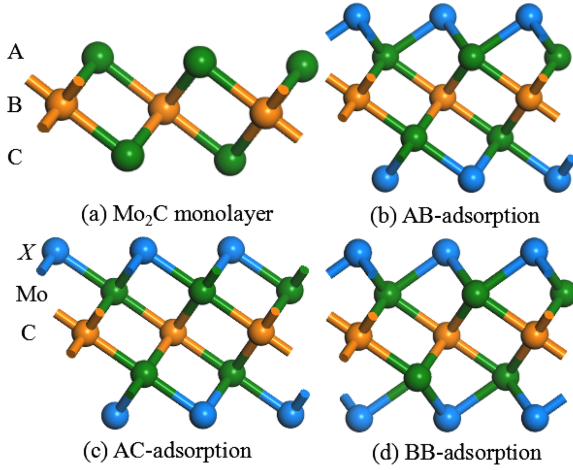


FIG. 1. Side views of atomic structures of monolayer Mo<sub>2</sub>C (a) and three atom-adsorbed configurations (b)-(d).

electronic wave function in the plane waves, whereas the cutoff energy for the charge density and potential is set to be 350 Ry. Structure optimization and electronic structure are repeated by using PWSCF, and the results are consistent with those obtained using VASP.

The vacuum space of  $\sim 15$  Å is intercalated into interlamination to eliminate the interaction between layers. A  $12 \times 12$  2D grid uniform is applied for both the  $k$ -points of the self-consistent and  $q$ -points of dynamical matrices calculations.

The EPC calculation is estimated according to the Migdal-Eliashberg theory [ $\alpha^2 F(\omega)$ ], which is given by<sup>21</sup>

$$\alpha^2 F(\omega) = \frac{1}{2\pi N(\epsilon_F)} \sum_{\mathbf{q}\nu} \delta(\omega - \omega_{\mathbf{q}\nu}) \frac{\gamma_{\mathbf{q}\nu}}{\hbar\omega_{\mathbf{q}\nu}}, \quad (1)$$

where  $N(\epsilon_F)$  is the electronic density of states (DOS) at the Fermi level,  $\omega_{\mathbf{q}\nu}$  denotes the phonon frequency of the  $\nu$ th phonon mode with wave vector  $\mathbf{q}$ , and the phonon linewidth  $\gamma_{\mathbf{q}\nu}$  is defined by<sup>22,23</sup>

$$\gamma_{\mathbf{q}\nu} = \frac{2\pi\omega_{\mathbf{q}\nu}}{\Omega_{BZ}} \sum_{ij} \int d^3k |g_{\mathbf{k}i, \mathbf{k}+\mathbf{q}j}^{\mathbf{q}\nu}|^2 \delta(\epsilon_{\mathbf{k}i} - \epsilon_F) \delta(\epsilon_{\mathbf{k}+\mathbf{q}j} - \epsilon_F), \quad (2)$$

where  $ij$  denotes the indices of energy bands,  $\Omega_{BZ}$  is the volume of Brillouin zone,  $\epsilon_{\mathbf{k}i}$  and  $\epsilon_{\mathbf{k}+\mathbf{q}j}$  are the eigenvalues of Kohn-Sham orbitals at given bands and wave vectors. The  $g_{\mathbf{k}, \mathbf{q}\nu}$  is the EPC matrix element which can be determined self-consistently by the linear response theory, which describes the probability amplitude for the scattering of an electron with a transfer of crystal momentum  $\mathbf{q}$ , and is determined by<sup>22,23</sup>

$$g_{\mathbf{k}, \mathbf{q}\nu}^{ij} = \left( \frac{\hbar}{2M\omega_{\mathbf{q}\nu}} \right)^{1/2} \langle \psi_{i, \mathbf{k}} | \frac{dV_{SCF}}{d\hat{u}_{\mathbf{q}\nu}} \cdot \hat{e}_{\mathbf{q}\nu} | \psi_{i, \mathbf{k}+\mathbf{q}} \rangle, \quad (3)$$

where  $M$  is the atomic mass,  $\frac{dV_{SCF}}{d\hat{u}_{\mathbf{q}\nu}}$  measures the change of self-consistent potential induced by atomic displacement,  $\psi_{i, \mathbf{k}}$  and  $\psi_{i, \mathbf{k}+\mathbf{q}}$  are the Kohn-Sham orbitals.

The EPC constant  $\lambda$  is obtained by summation over the first Brillouin zone or integration of the  $\alpha^2 F(\omega)$  in the  $\mathbf{q}$  space<sup>22,23</sup>

$$\lambda = \sum_{\mathbf{q}\nu} \lambda_{\mathbf{q}\nu} = 2 \int_0^\infty \frac{\alpha^2 F(\omega)}{\omega} d\omega, \quad (4)$$

where the EPC constant  $\lambda_{\mathbf{q}\nu}$  for mode  $\nu$  at wave vector  $\mathbf{q}$  is defined by the integration<sup>22,23</sup>

$$\lambda_{\mathbf{q}\nu} = \frac{\gamma_{\mathbf{q}\nu}}{\pi\hbar N(\epsilon_F)\omega_{\mathbf{q}\nu}^2}. \quad (5)$$

To obtain accurate electron-phonon interaction matrices, a dense  $36 \times 36 \times 1$  grid is adopted for the EPC calculation.

### III. RESULTS AND DISCUSSION

#### A. Unfunctionalized monolayer Mo<sub>2</sub>C

The structure of unfunctionalized monolayer Mo<sub>2</sub>C is layered hexagonal with a space group of  $D_{3d}$ , and the stacking of Mo–C–Mo is in the ABC-type along the hexagonal  $c$  axis (see Fig. 1(a)), which is similar to 1T-MoS<sub>2</sub>. Our optimized lattice constant (2.965 Å) is consistent with the value reported by Khazaei *et al.*<sup>10</sup>

The calculated band structures and density of states (DOS) are shown in Fig. 2. As previously reported,<sup>10</sup> monolayer Mo<sub>2</sub>C indicates a strong metallic behavior, and the Fermi surfaces are mainly contributed by Mo's  $d$ -orbitals according to the projected DOS (Fig. 2(a)). The maximally localized Wannier functions (MLWFs) can partition Mo's  $d$  orbital and C's  $p$  orbital, as shown in Fig. 2(b). C's  $p$  orbital is mainly occupied in the energy range from  $-8$  eV to  $-4$  eV and highly hybridize with Mo's  $d$  orbitals. The  $p_x$  and  $p_y$  orbitals are degenerate, which are higher in energy than the  $p_z$  orbital, as shown in Fig. 2(a). The Mo's  $d$ -orbital is split due to the  $C_{3v}$  crystalline field. Thus the  $d_{z^2}$  has the lowest on-site energy, the rest are pairwise degenerate: ( $d_{x^2-y^2}$  and  $d_{xy}$ ) and ( $d_{xz}$  and  $d_{yz}$ ), the latter of which is slightly higher. The SOC can open the degeneracy at  $\bar{\Gamma}$  point (see Fig. 2(c)), while it has little effect on the Fermi surface. The crossing-Fermi-level bands and corresponding Fermi surfaces are shown in Figs. 2(d) and 2(e). The antibonding  $d_{z^2}$  band (the upper one in Fig. 2(e)) forms the hole pocket Fermi surfaces around  $\bar{M}$  points, while the electron Fermi surface (the lower one in Fig. 2(e)) is the circular-shape around the  $\bar{\Gamma}$  point which is mainly contributed by  $d_{x^2-y^2}$  and  $d_{xy}$ . Such double Fermi surfaces correspond to carriers with multiple effective masses, charges, and mobilities.

Then the phonon properties and electron-phonon coupling are calculated. The SOC is not included in these calculations, considering its negligible effects on the Fermi surfaces. For monolayer Mo<sub>2</sub>C, the Raman modes can be decomposed as  $A_{1g}^1 \oplus 2E_g^1$  at the zone center (see Fig. 3(a)), and the calculated Raman frequencies are  $150.3$  cm<sup>-1</sup> and  $193.1$  cm<sup>-1</sup> for the  $E_g^1$  and  $A_{1g}^1$  modes, respectively. Moreover, both the Raman modes have strong coupling to electrons ( $\lambda_{\mathbf{q}\nu}$  is 0.11 for  $E_g^1$  and 0.21 for  $A_{1g}^1$ ) according to Eq. (5).

The calculated phonon dispersions along major high symmetry lines and phonon densities of states (PDOS,  $F(\omega)$ ) for monolayer Mo<sub>2</sub>C are shown in Fig. 3(a). More dense  $k$ -meshes

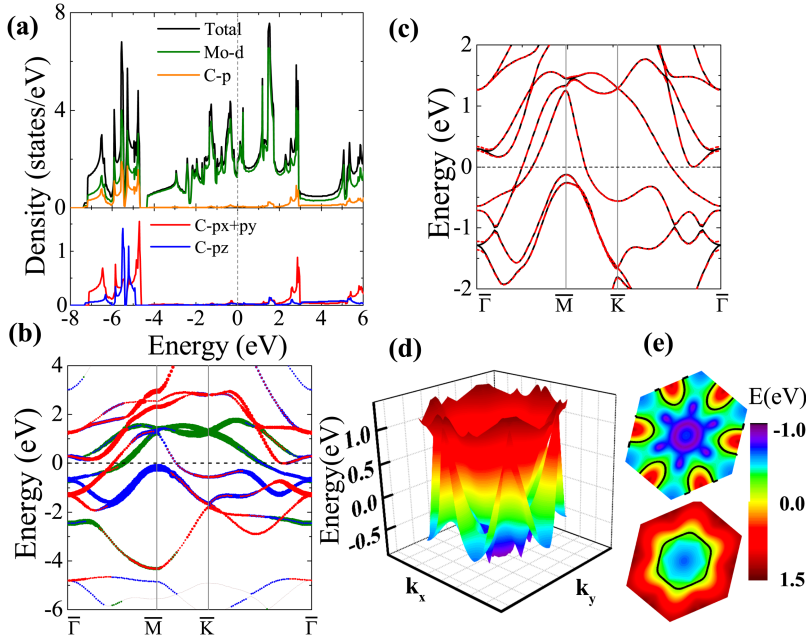


FIG. 2. Electronic structure of monolayer  $\text{Mo}_2\text{C}$ . (a) Density of states (DOS) and Projected DOS. (b) Projected band structure. Red:  $d_{z^2}$ ; green:  $d_{x^2-y^2}$  and  $d_{xy}$ ; blue:  $d_{xz}$  and  $d_{yz}$ . (c) Band structure with (black) and without SOC (red). (d) Three-dimensional view of two bands cross the Fermi level. (e) Energy counterplots of two bands. Black curves: the Fermi surface.

( $18 \times 18 \times 1$  and  $24 \times 24 \times 1$ ) in self-consistent calculations are also tested. The maximum error of obtained phonon frequencies are less than 1%, implying the convergence of the phonon calculation. In addition, the different methodology and pseudo-potential would lead to a few differences regarding the phonon band structures.<sup>8,11</sup> In fact, our phonon structure is very close to that in previous report,<sup>11</sup> although tiny differences remain unavoidable.

No imaginary frequency exists in the full phonon spectra, indicating the dynamical stability of monolayer  $\text{Mo}_2\text{C}$ . Meanwhile, the phonon behavior exhibits several remarkable characteristics. First, near the zone center, both the LA and TA branches are near linear while the ZA branch (out-of-plane acoustical mode) is quadratic. These characters reflect the nature of the 2D sheet. In detail, the ZA phonon in 2D materials like graphene has a quadratic dispersion over a wide range of the 2D Brillouin zone  $\omega_{\text{ZA}} = a_{\text{ZA}}q^2$ , where  $a_{\text{ZA}}$  is a positive constant and  $q$  is the 2D phonon wave vector. The similar conclusions are also found in monolayer black phosphorene.<sup>24</sup> Second, according to the partial PDOS (Fig. 3(b)), the vibrational modes of Mo dominate the low-frequency regime while those of C dominate the high-frequency regime due

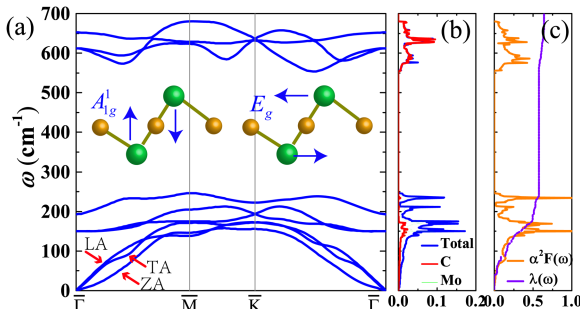


FIG. 3. Phonon properties for a single  $\text{Mo}_2\text{C}$  layer. (a) Calculated phonon dispersion. Inset: sketch of Raman modes. ((b) and (c)) Phonon DOS, electron-phonon coupling  $\lambda$ , and Eliashberg spectral function.

to their large difference in mass. There is a large gap around  $300 \text{ cm}^{-1}$ , which partitions the optical modes of Mo and C.

To discuss the superconductivity, the obtained  $\alpha^2F(\omega)$  and  $\lambda(\omega)$  are also plotted in Fig. 3(c). Their similar shapes indicate that all  $F(\omega)$  make contributions to EPC. Due to the factor  $1/\omega$  in the definition of  $\lambda$  (see Eq. (4)), the contributions from the low  $\omega$  region is more prominent. Exactly, the calculated  $\lambda(\omega = 250 \text{ cm}^{-1})$  is  $\approx 0.56$  which is beyond 90% of the total EPC ( $\lambda(\omega = \infty) = 0.63$ ), indicating that the phonon modes in the frequency region below  $250 \text{ cm}^{-1}$  have the dominant contribution. In particular, three low-lying optical branches have strong coupling to electrons, which make 40% contribution to EPC. Therefore, it is natural to expect the strong EPC in monolayer  $\text{Mo}_2\text{C}$  to induce the superconducting state. The  $T_C$  can be estimated using the Allen-Dynes modified McMillan equation,<sup>22</sup>

$$T_C = \frac{\omega_{\text{ln}}}{1.2} \exp\left[-\frac{1.04(1 + \lambda)}{\lambda - \mu^*(1 + 0.62\lambda)}\right], \quad (6)$$

where  $\mu^*$  is the Coulomb repulsion parameter and  $\omega_{\text{ln}}$  is the logarithmically averaged frequency. When taking a typical value  $\mu^* = 0.1$ , the estimated  $T_C$  is about 5.9 K.

TABLE I. The calculated total energy for  $\text{Mo}_2\text{CX}_2$  of different configurations for adsorption is in unit of eV/per u.c. The energy of BB-adsorption is set as the reference. The optimized lattice constant  $a$  is in unit of Å. For the most favorable configuration, the corresponding binding energy ( $E_b$ ) (in unit of eV) is also presented.

	AB	AC	BB	$a$	$a$ (Ref. 10)	$E_b$
O	0.72	1.29	0	2.891	2.886	-8.67
S	0.48	0.69	0	3.087	3.078	-6.08
Se	0.39	0.24	0	3.161		-4.57
Br	-0.19	-0.65	0	3.428	3.418	-3.13

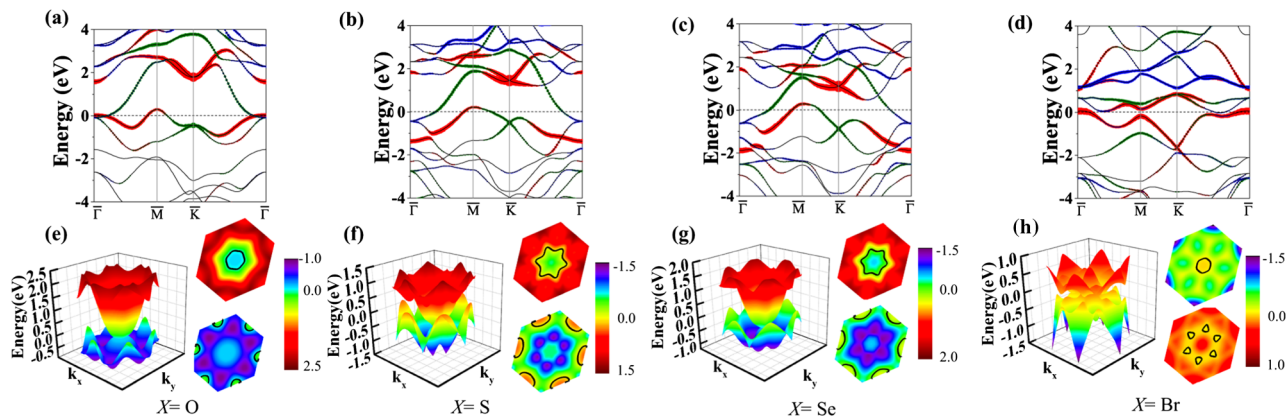


FIG. 4. Electronic structures of  $\text{Mo}_2\text{CX}_2$ . ((a)-(d)) Band structures. Red:  $d_{z^2}$ ; green:  $d_{x^2-y^2}$  and  $d_{xy}$ ; blue:  $d_{xz}$  and  $d_{yz}$ . ((e)-(h)) Three-dimensional view of three bands around the Fermi level and the corresponding Fermi surface (black curves).

## B. Functionalized monolayer $\text{Mo}_2\text{C}$

In the above study, it has been predicted that pure monolayer  $\text{Mo}_2\text{C}$  maybe a quasi-2D superconductor. However, in real situations, the functional group at surfaces of monolayer  $\text{Mo}_2\text{C}$  is unavoidable considering its highly unsaturated surfaces. In fact, previous studies reported that monolayer  $\text{Mo}_2\text{C}$  transforms from metal to semiconductor with F- and Cl-adsorption,<sup>10</sup> namely, the F- and Cl-functionalized monolayer  $\text{Mo}_2\text{C}$  could not be a superconductor. In the following, the changes of superconductivity by adsorbing various functional groups will be studied, which may shed light to tuning the superconductivity of monolayer  $\text{Mo}_2\text{C}$  in real experiments.

As the full adsorption is more stable than the partial case,<sup>10</sup> the  $1 \times 1$   $\text{Mo}_2\text{C}$  unit cell (u.c.) with two functional groups  $X$  (one on each surface), i.e.,  $\text{Mo}_2\text{CX}_2$ , is adopted in our calculation. Four functional atoms ( $X = \text{O}$ ,  $\text{S}$ ,  $\text{Se}$ , and  $\text{Br}$ ) are considered. Considering the symmetry, there are three mostly possible site-configuration for adsorption: AB-adsorption (Fig. 1(b)), AC-adsorption (Fig. 1(c)), and BB-adsorption

(Fig. 1(d)). For the AB-adsorption, one  $X$  atom is right above the C layer, while another  $X$  is right below other side Mo. For the AC-adsorption, each  $X$  atom is right above/below the other side Mo layer. For the BB-configuration, both two  $X$  atoms stand above/below the C site.

The crystal structures are fully relaxed upon the absorptions, and the calculated lattice constant and total energy are listed in Table I. Our calculations are in good agreement with previous reports,<sup>10,25</sup> and the BB-adsorption is the most favorable case for chalcogen, while  $\text{Mo}_2\text{CBr}_2$  favors the AC-adsorption. The corresponding binding energy ( $E_b = E_{\text{Mo}_2\text{CX}_2} - E_{\text{Mo}_2\text{C}} - E_{X_2}$ ) is also calculated and also listed in Table I. All binding energies are negative which indicate thermodynamic stability for all structures. And the value of binding energy decreases from  $\text{Mo}_2\text{CO}_2$  to  $\text{Mo}_2\text{CBr}_2$ , implying the adsorption of oxygen should be quite possible, as observed in the real experiment.<sup>9</sup> According to the Löwdin population analysis, the charge transfer from Mo to  $X$  is about 0.48, 0.39, 0.34, and 0.24 electron for O-, S-, Se-, and Br-adsorption, respectively. The stronger Coulomb attraction between  $X$  and Mo ions can lead to larger binding energy.

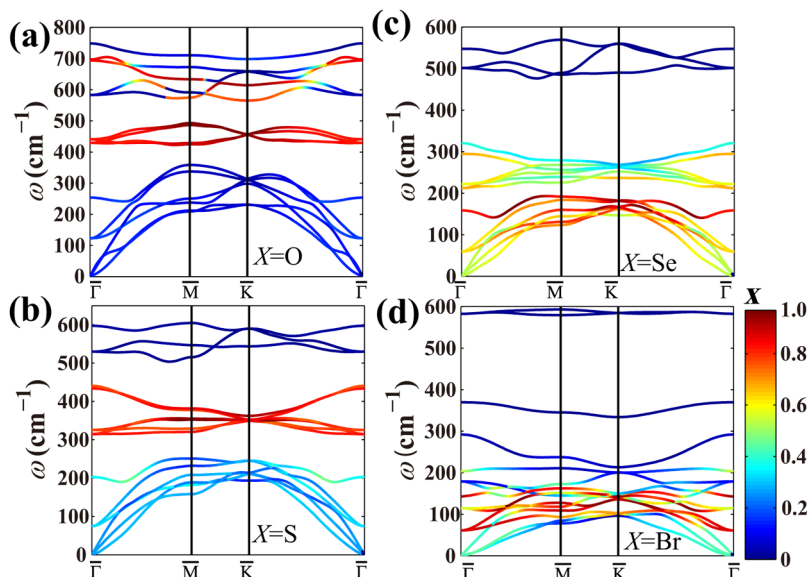


FIG. 5. Calculated phonon dispersion for  $\text{Mo}_2\text{CX}_2$ . The contribution from  $X$  and monolayer  $\text{Mo}_2\text{C}$  is distinguished by color.

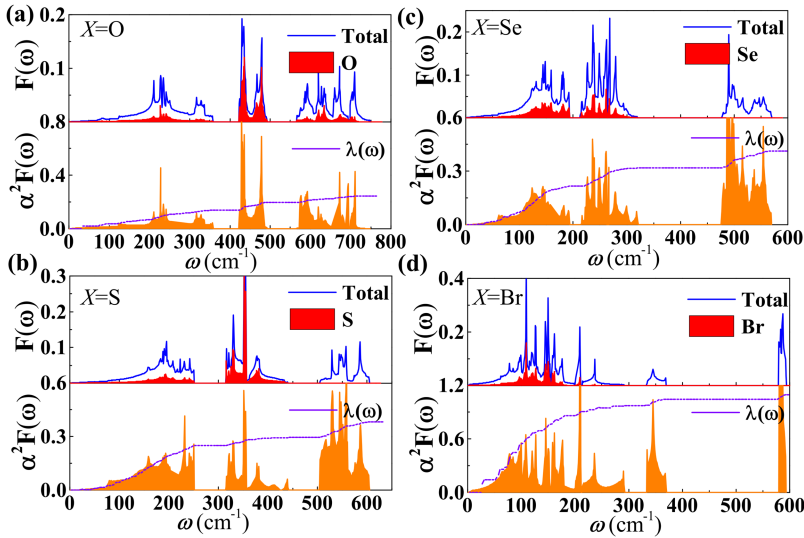


FIG. 6. Phonon DOS (PDOS,  $F(\omega)$ ), projected PDOS of  $X$  atoms, electron-phonon coupling  $\lambda(\omega)$ , and Eliashberg spectral function of  $\text{Mo}_2\text{CX}_2$ .

The electronic structures for  $\text{Mo}_2\text{CX}_2$  are calculated. Due to the unchanged symmetry of the crystalline field, the splitting of Mo's  $d$  orbitals is similar to pristine monolayer  $\text{Mo}_2\text{C}$ . Quantitatively, the absorption of  $X$  further raises the energy of doubly degenerate  $d_{xz}$  and  $d_{yz}$  due to the strong hybridization with the  $p$  orbitals of  $X$ . In detail, the on-site energy difference between  $d_{xz}/d_{yz}$  and  $d_{x^2-y^2}/d_{xy}$  is about 1.72, 0.99, 0.71, and 0.21 eV for the O-, S-, Se- and Br-adsorption, respectively, according to the MLWFs calculation. The lower of  $d_{xz}/d_{yz}$  orbitals can be also evidenced in the band structures of  $\text{Mo}_2\text{CX}_2$ , as shown in Figs. 4(a)–4(d). All band structures show a metallic character after  $X$ -adsorption, different to previously studied F-/Cl-adsorption. Similar to pristine monolayer  $\text{Mo}_2\text{C}$ ,  $d_{x^2-y^2}/d_{xy}$  and  $d_{z^2}$  of Mo still make dominant contributions around the Fermi level, although the contributions from  $d_{x^2-y^2}/d_{xy}$  are reduced more or less, as shown in Figs. 4(a)–4(d). The three-dimensional view of three bands as well as the Fermi surface of  $d_{x^2-y^2}/d_{xy}$  and  $d_{z^2}$  is shown in Figs. 4(e)–4(h). For chalcogen, its band structure looks like that of pristine monolayer  $\text{Mo}_2\text{C}$ , namely, the upper cone-shape band is surrounded by a lower flower-shape band, although the degree of surrounding is suppressed. Due to the different adsorption site and valence state of Br, the shapes of Fermi surfaces are significantly changed to sun-like patterns for  $\text{Mo}_2\text{CBr}_2$ .

The calculated phonon dispersion along major high symmetry lines is shown in Fig. 5. No imaginary frequency exists in the full phonon spectra, indicating the dynamical stability of  $\text{Mo}_2\text{CX}_2$ . Therefore, it is more likely that  $\text{Mo}_2\text{CX}_2$  is able to be obtained in the real experiment considering its stability in thermodynamics and lattice dynamical. Obviously, the vibration modes of  $\text{Mo}_2\text{C}$  are strongly coupled with the surface  $X$ , as revealed by the phonon dispersion (Fig. 5). Different  $X$  atoms contribute to the phonon spectrum in different frequency ranges due to the inequivalent mass and bond strength. In particular, for  $\text{Mo}_2\text{CO}_2$ , the contributions from O mainly locate at the intermediate- and high-frequency regimes, indicating the fact of the strong bond of Mo–O. In the case of  $\text{Mo}_2\text{CS}_2$ , the frequencies contributed by S mainly locate in the intermediate regime, while for Se and

Br cases the contributions from  $X$  are in the low frequency side.

Our results for  $\alpha^2 F(\omega)$ ,  $F(\omega)$ , and PDOS of  $\text{Mo}_2\text{CX}_2$  are shown in Fig. 6. As in the monolayer  $\text{Mo}_2\text{C}$  case,  $\alpha^2 F(\omega)$  and  $F(\omega)$  of  $\text{Mo}_2\text{CX}_2$  have similar peaks, indicating all vibration modes contribute to EPC. Comparing to pristine  $\text{Mo}_2\text{C}$ , the strength of  $\alpha^2 F(\omega)$  has been suppressed by O-, S-, and Se-adsorption. And such suppression obviously exists in the low frequency regime which has large contributions to EPC (because of the  $\omega^{-1}$  part in Eq. (4)). The average EPC is also calculated using Eq. (4), as listed in Table II. Corresponding  $T_C$  is estimated from the modified McMillan equation (Eq. (6)) with  $\mu^* = 0.1$  (listed in Table II). The results indicate that the superconductivity is greatly suppressed in  $\text{Mo}_2\text{CS}_2$  and  $\text{Mo}_2\text{CSe}_2$  and almost disappears in  $\text{Mo}_2\text{CO}_2$ , as a result of reduced electronic DOS's at the Fermi level and suppressed EPC's.

As shown in Fig. 6(d), the strength of  $\alpha^2 F(\omega)$  has been improved in  $\text{Mo}_2\text{CBr}_2$ , and the corresponding average EPC is about 1.09. The obtained  $\lambda(\omega = 300 \text{ cm}^{-1}) \approx 0.97$  is approximately beyond 88% of the total EPC ( $\lambda(\omega = \infty) = 1.09$ ), and the estimated  $T_C$  is up to 12.8 K. To be exact, EPC of vibration modes ( $E_g^1$  and  $A_{1g}^1$ ) at the  $\Gamma$  point which have Raman activity have been improved for 30% comparing to those of pristine monolayer  $\text{Mo}_2\text{C}$ . In addition, the large values of EPC for acoustic modes at  $\bar{M}$  contribute substantially to the average EPC. Thus, the superconductive  $T_C$  is pushed up in  $\text{Mo}_2\text{CBr}_2$ .

In addition, we had tried the full-OH cover up  $\text{Mo}_2\text{C}$  surfaces. However, even after the atomic relaxation, an imaginary frequency of phonon appears at the  $\bar{M}$  point, implying an

TABLE II. The calculated  $\text{Mo}_2\text{CX}_2$ 's superconductive parameters of  $N(\varepsilon_F)$  (states/eV),  $\omega_m$  (K),  $\lambda$ , and  $T_C$  (K).

	$\text{Mo}_2\text{CO}_2$	$\text{Mo}_2\text{CS}_2$	$\text{Mo}_2\text{CSe}_2$	$\text{Mo}_2\text{CBr}_2$
$N(\varepsilon_F)$	1.3	1.5	1.6	3.3
$\omega_m$	357.4	326.6	283.7	160.7
$\lambda$	0.2	0.4	0.4	1.1
$T_C$	<0.1	1.0	1.4	12.8

unstable structure. Therefore, the structural phase transition would appear which makes the problem more complicated. Thus, the decoration of OH group is beyond the current work and deserves individual studies in future.

#### IV. CONCLUSION

We have analyzed the electronic properties, the lattice dynamical properties, and the superconductivity of monolayer  $\text{Mo}_2\text{C}$  and its functionalized ones. Our calculations have confirmed the strong EPC in monolayer  $\text{Mo}_2\text{C}$ , which may lead to superconductivity below 5.9 K. Even though, since the absorption of functional groups is unavoidable in the real experiment, its superconductivity can be modified. Our calculation has found that for chalcogen functionalized monolayer  $\text{Mo}_2\text{C}$  the superconductivity would be seriously suppressed (or even totally disappear). The most interesting prediction is that the electron-phonon coupling can be greatly enhanced in monolayer  $\text{Mo}_2\text{C}$  by bromine absorption, and thus its corresponding superconductive  $T_C$  may be pushed up to 12.8 K, suggesting that  $\text{Mo}_2\text{CBr}_2$  may be a good candidate as a nanoscale superconductor.

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<sup>1</sup>M. Naguib, M. Kurtoglu, V. Presser, J. Lu, J. Niu, M. Heon, L. Hultman, Y. Gogotsi, and M. W. Barsoum, *Adv. Mater.* **23**, 4248 (2011).

<sup>2</sup>M. Naguib, O. Mashtalir, J. Carle, V. Presser, J. Lu, L. Hultman, Y. Gogotsi, and M. W. Barsoum, *ACS Nano* **6**, 1322 (2012).

- <sup>3</sup>M. Naguib, J. Halim, J. Lu, K. M. Cook, L. Hultman, Y. Gogotsi, and M. W. Barsoum, *J. Am. Chem. Soc.* **135**, 15966 (2013).
- <sup>4</sup>M. R. Lukatskaya, O. Mashtalir, C. E. Ren, Y. Dall'Agnese, P. Rozier, P. L. Taberna, M. Naguib, P. Simon, M. W. Barsoum, and Y. Gogotsi, *Science* **341**, 1502 (2013).
- <sup>5</sup>S. Zhao, W. Kang, and J. Xue, *Appl. Phys. Lett.* **104**, 133106 (2014).
- <sup>6</sup>M. Khazaei, M. Arai, T. Sasaki, C.-Y. Chung, N. S. Venkataramanan, M. Estili, Y. Sakka, and Y. Kawazoe, *Adv. Funct. Mater.* **23**, 2185 (2013).
- <sup>7</sup>M. Ghidui, M. R. Lukatskaya, M.-Q. Zhao, Y. Gogotsi, and M. W. Barsoum, *Nature* **516**, 78 (2014).
- <sup>8</sup>W. Sun, Y. Li, B. Wang, X. Jiang, M. I. Katsnelson, P. Korzhavyi, O. Eriksson, and I. Di Marco, *Nanoscale* **8**, 15753 (2016).
- <sup>9</sup>R. Meshkian, L.-Å. Näslund, J. Halim, J. Lu, M. W. Barsoum, and J. Rosen, *Scr. Mater.* **108**, 147 (2015).
- <sup>10</sup>M. Khazaei, M. Arai, T. Sasaki, M. Estili, and Y. Sakka, *Phys. Chem. Chem. Phys.* **16**, 7841 (2014).
- <sup>11</sup>X.-H. Zha, J. Yin, Y. Zhou, Q. Huang, K. Luo, J. Lang, J. S. Francisco, J. He, and S. Du, *J. Phys. Chem. C* **120**, 15082 (2016).
- <sup>12</sup>J.-F. Ge, Z.-L. Liu, C. Liu, C.-L. Gao, D. Qian, Q.-K. Xue, Y. Liu, and J.-F. Jia, *Nat. Mater.* **14**, 285 (2015).
- <sup>13</sup>J.-J. Zhang, B. Gao, and S. Dong, *Phys. Rev. B* **93**, 155430 (2016).
- <sup>14</sup>J.-J. Zhang and S. Dong, *2D Mater.* **3**, 035006 (2016).
- <sup>15</sup>C. Xu, L. Wang, Z. Liu, L. Chen, J. Guo, N. Kang, X.-L. Ma, H.-M. Cheng, and W. Ren, *Nat. Mater.* **14**, 1135 (2015).
- <sup>16</sup>M. K. Kolel-Veetil, S. B. Qadri, M. Osofsky, T. M. Keller, R. Goswami, and S. A. Wolf, *J. Phys. Chem. C* **111**, 16878 (2007).
- <sup>17</sup>Y. Xie and P. R. C. Kent, *Phys. Rev. B* **87**, 235441 (2013).
- <sup>18</sup>G. Kresse and J. Furthmüller, *Phys. Rev. B* **54**, 11169 (1996).
- <sup>19</sup>G. Kresse and D. Joubert, *Phys. Rev. B* **59**, 1758 (1999).
- <sup>20</sup>P. Giannozzi, S. Baroni, N. Bonini, M. Calandra, R. Car, C. Cavazzoni, D. Ceresoli, G. L. Chiarotti, M. Cococcioni, I. Dabo *et al.*, *J. Phys.: Condens. Matter* **21**, 395502 (2009).
- <sup>21</sup>G. Grimvall, *The Electron-Phonon Interaction in Metals*, Vol. 8 (North-Holland, Amsterdam, 1981).
- <sup>22</sup>P. B. Allen and R. C. Dynes, *Phys. Rev. B* **12**, 905 (1975).
- <sup>23</sup>P. Allen, *Phys. Rev. B* **6**, 2577 (1972).
- <sup>24</sup>G. Qin, Q.-B. Yan, Z. Qin, S.-Y. Yue, M. Hu, and G. Su, *Phys. Chem. Chem. Phys.* **17**, 4854 (2015).
- <sup>25</sup>H. Weng, A. Ranjbar, Y. Liang, Z. Song, M. Khazaei, S. Yunoki, M. Arai, Y. Kawazoe, Z. Fang, and X. Dai, *Phys. Rev. B* **92**, 075436 (2015).