Sliding-mediated ferroelectric phase transition in CuInP₂S₆ under pressure **5**

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ABSTRACT

Interlayer stacking order has recently emerged as a unique degree of freedom to control crystal symmetry and physical properties in twodimensional van der Waals (vdW) materials and heterostructures. By tuning the layer stacking pattern, symmetry-breaking and electric polarization can be created in otherwise non-polar crystals, whose polarization reversal depends on the interlayer sliding motion. Herein, we demonstrate that in a vdW layered ferroelectric, its existing polarization is closely coupled to the interlayer sliding driven by hydrostatic pressure. Through combined structural, electrical, vibrational characterizations, and theoretical calculations, we clearly map out the structural evolution of $CuInP_2S_6$ under pressure. A tendency toward a high polarization state is observed in the low-pressure region, followed by an interlayer-sliding-mediated phase transition from a monoclinic to a trigonal phase. Along the transformation pathway, the displacive-instable Cu ion serves as a pivot point that regulates the interlayer interaction in response to external pressure. The rich phase diagram of $CuInP_2S_6$, which is enabled by stacking orders, sheds light on the physics of vdW ferroelectricity and opens an alternative route to tailoring long-range order in vdW layered crystals.

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INTRODUCTION

Recent developments in two-dimensional van der Waals (vdW) ferroelectric materials have brought unexpected twists to ferroelectric physics.^{1,2} The reduced lattice dimensionality and large degrees of freedom for layer stacking enable unconventional properties in two-dimensional (2D) ferroelectrics in comparison to their three-dimensional (3D) counterparts, such as sliding/moiré ferroelectricity^{3–6} and layer-dependent cumulative polarization.^{7,8} As a classic vdW layered crystal with robust room-temperature ferroelectricity, CuInP₂S₆ (CIPS) has received revived interest recently, due to its

fascinating ferroelectric characteristics.^{9,10} Most notable examples include negative longitudinal piezoelectricity,^{11,12} strongly intertwined polarization switching and ionic conductivity,^{13,14} as well as quadruple-well polarization states.¹⁵ All these unique features stem from the displacive instability of the monovalent Cu⁺ in the 2D crystal lattice. Hence, thorough investigations of the displacement behavior of Cu⁺ in CIPS under electric, strain, and temperature stimuli hold great importance for understanding the ferroelectric physics in this intriguing compound and may deliver more general insights into relevant 2D ferroelectrics.

One particular point of interest is the metastable Cu site within the vdW gap, which lies at the core of the observed negative d₃₃ and strain-tunable quadruple-well ferroelectricity. Previous work demonstrated that it is possible to stabilize the occupancy of the in-gap Cu site by heterostructure strain in a mixed-phase CuInP2S6-In4/3P2S6 solid solution, resulting in a high-polarization state with inverse piezoelectric sign.¹⁵ However, the inhomogeneous composition and strain distribution across the sample complicates the intrinsic property characterizations, and one may naturally beg the question: can we realize the high polarization state by applying a homogeneous hydrostatic stress? To this end, it is necessary to study how phase pure CIPS behaves under high pressure. Unlike conventional positive-d₃₃ ferroelectrics, whose polarization decreases and, thus, ferroelectric order deteriorates on the application of pressure,¹⁶ CIPS should behave oppositely.¹⁷ In fact, early high-pressure work on CIPS discovered a first-order phase transition at \approx 4.0 GPa,¹⁸ with a potential symmetry change from monoclinic to trigonal. Later, the dependence of dielectric properties of CIPS on application of pressure was studied,^{19,20} which revealed an enhancement of the Curie temperature (T_C) as expected for a ferroelectric with negative piezoelectricity. However, detailed structural and polarization evolutions of CIPS under hydrostatic pressure remain unclear. Herein, we have studied the effect of pressure on CIPS using single-crystal high-pressure x-ray diffraction (XRD), electrical conductivity, Raman spectroscopy, and theoretical calculations, in order to map out the structural evolution of CuInP2S6 and its subsequent physical properties.

RESULTS

Structural evolution of CIPS under hydrostatic pressure

CIPS crystallizes in a vdW monolayered structure, the structural backbone of which is an edge-shared S₆ octahedral framework, inside which Cu, In ions, and P-P pairs fill in order, forming a triangular lattice. The second layer stacks atop by first rotating 180° and shifting by $1/3 \,\overline{a}$ along the *a*-axis, resulting in monoclinic lattice symmetry (space group Cc) with \overline{b} as the unique axis.²¹ Displacement of the Cu atom from the octahedral center is the main source of ferroelectricity for CIPS. With increasing temperature, the Cu ion exhibits strong thermal broadening perpendicular to the basal plane.²² Both experimental and theoretical studies have shown that this thermal hopping effect is accompanied by an additional metastable site within the vdW gap,¹⁷ analogous to the second-order Jahn Teller effect found in zinc blende CuCl.²³ Therefore, the occupancy of the Cu is presented as two independent sites [Fig. 1(a)], of which one is located within the S_6 octahedron, and referred to as "inside Cu" (Cu^I), whereas the other Cu-site, sits within the vdW gap ["outside Cu" (Cu $^{\rm O}$)]. In other words, the Cu $^{\rm O}$ site is stabilized by the interlayer interaction with the S atom across the vdW gap.

The structural evolution of CIPS on increasing pressures at room temperature is shown in Fig. 1(a), as obtained from single-crystal XRD measurements. At the early stage of the pressurization, the basic crystal structure of CIPS remains unchanged. However, the relative population of the Cu ions between the two sites varies rapidly on increasing pressure. Particularly, the fraction of the Cu^O site increases continuously, from 10% under ambient pressure to $\approx 60\%$ at ≈ 1 GPa [Fig. 1(e)]. On increasing pressure further, the Cu occupancies stay insensitive to the pressure change. At ≈ 4 GPa, the pressure-induced structural phase transition occurs. The adjacent layers slide by 1/3 \vec{a} along the *a*-axis, transforming from monoclinic to trigonal symmetry (space group $P\bar{3}1c$). Due to the interlayer sliding, the Cu column is no longer aligned with the S4 atom of the adjacent layer [labeled S4 in Fig. 1(a)], but in line with the P-P column. Consequently, the stacking sequence of the sulfur layers switches from ABC-type to AB-type closed pack (Fig. S1). Meanwhile, the Cu ions jump back into the sulfur octahedral cage, with the majority staying at the central site and the rest distributing symmetrically between the upper and lower Cu^I sites. As a result, CIPS now becomes a centrosymmetric non-polar phase.

On increasing pressure to \approx 3.5 GPa, and prior to the phase transition, CIPS compresses hydrostatically, with a gradual reduction in cell volume [Fig. 1(c)]. The compressibility within the plane of the intralayer is smaller than that of the c^* axis (normal to the plane), whose length is reduced by 5% up to \approx 3.5 GPa [Fig. 1(b)]. This is consistent with the highly anisotropic nature of chemical connectivity, as a result of the vdW layered structure. At the phase transition, there is a sudden contraction of \approx 2% along the c^* direction which is accompanied by interlayer sliding. The interlayer bond between Cu and S4 is also broken due to the sliding (average bond length changes from 2.63 to 3.29 Å), resulting in further collapse of the vdW gap thickness [Fig. 1(e)]. Concomitantly, the cell volume also exhibits a discontinuous jump at the phase transition boundary [Fig. 1(c)]. The bulk modulus K₀ is estimated to be 25.2(3.3) GPa, consistent with previous report.²⁴ The monoclinic angle β also increases slightly on increasing pressure [Fig. 1(d)], as a result of the weak sliding between adjacent layers. By scrutinizing the evolutions of layer and vdW gap thicknesses (defined by the average z position of the sulfur planes), how the structure of CIPS responds to compression can be followed [Fig. 1(e)]. The initial drop of layer thickness and rise of the gap thickness are in accordance with the drastic relocation of Cu^I toward Cu^O sites at lower pressures [Fig. 1(f)]. Further pressurization barely affects the layer thickness, but compresses the gap between the layers. On undergoing the phase transition, there is a sudden reduction in the gap thickness, consistent with the change in the lattice parameter along *c*-axis. On the contrary, the layer thickness expands in the trigonal phase, because all the Cu ions now jump back into the octahedral cage. As an indicator of the polarization magnitude, the average off-center displacement of Cu ion with respect to the octahedral center is also derived, which shows a similar trend as the pressure-dependent Cu^O occupancy [Fig. 1(g)]. The enhanced off-center displacement of Cu ion also results in strong puckering of the sulfur plane, due to the contraction of the Cu-S4 bond [Fig. S2(b)]. Structural distortions of the sulfur octahedra also takes place, as manifested by the expansion and contraction of the upper and lower S3 triangle in the CuS6 sublattice as well as the counter-rotations of the In and P-P octahedra [Fig. S2(e)]. These structural distortions are visibly suppressed in the high-pressure trigonal phase [Figs. S2(c) and S2(f)]. Previous theoretical calculation also confirmed a small but nonzero in-plane polarization component (P_x) in CIPS along the a-axis.²⁵ Structurally, P_x is associated with the relative in-plane displacement (\boldsymbol{x}_{Cu-S4}) between Cu and the S4 atom. Under pressurization, owing to the interlayer-sliding, x_{Cu-S4} gradually decreases, suggesting the reduction of P_x (Fig. S3).

To further understand above experimental discovery, we performed first-principles density functional theory (DFT) calculations. The possible phase transition caused by pressure in CIPS is determined through a comparison of enthalpies of Cc and $P\bar{3}1c$ phase. The enthalpy as function of pressure is calculated based on H = E + PV,



FIG. 1. Structural evolution of CIPS under hydrostatic pressure. (a) Upper panel: the evolution of the crystal structure of CIPS with hydrostatic pressure. The pink, purple, and yellow balls represent In, P, and S atoms, respectively. The blue–white balls are Cu sites and the proportion of blue colors indicates the corresponding occupancy of the respective site. Lower panel: zoomed-in image of the CuS_6 sublattice. The ball size of the Cu represents the corresponding occupancy of the site. (b) The lattice parameters, (c) cell volume, (d) monoclinic angle β , (e) layer and gap thicknesses, (f) the fraction of Cu¹ and Cu⁰ sites, and (g) average off-center displacement of Cu ion as a function of the applied hydrostatic pressure.

where E is the total DFT energy, P is pressure, and V is volume of unit cell. Our calculation indeed finds that CIPS undergoes a pressureinduced phase transition at \approx 5.8 GPa (at 0 K), from monoclinic (space group *Cc*) to trigonal phase (space group *P*31*c*) [Fig. S4(c)]. In the calculation, the *P*31*c* phase is the high-symmetry paraelectric phase with Cu lying at the center of the S₆ octahedron. However, DFT results reveal that the *P*31*c* phase exhibits one unstable phonon mode at Γ point, which is mainly contributed from Cu displacements, consequently leading to spontaneous phase transition from paraelectric $P\bar{3}1c$ to polar P31c structure [Fig. S4(b)]. The calculated energy difference between the P31c and $P\bar{3}1c$ phase is only 28.7 meV/f.u. Thus, it is clear that the ground state (at 0 K) of CIPS under high pressure is a polar ferroelectric, while it turns to be the paraelectric one at room temperature as experimentally observed. A rough estimation of the ferroelectric phase transition temperature is 51 K (see Discussion for details). The critical pressure in our calculation is close but slightly higher than the experimental one, which is also reasonable considering the thermal effect in room temperature experiments.

Our DFT calculation also shows that the Cu atoms occupy the Cu^{I} site below 0.5 GPa, before preferring the Cu^{O} site above 1 GPa, a similar trend of Cu displacement (compared to the average Cu position) from the experimental data [Fig. S4(d)]. However, thermal hopping of the Cu ion causes a mixed occupancy between the Cu^I and Cu^O sites, leading to a broadening of its distribution and smaller displacement of the center of mass observed at room temperature.

Electrical measurements under hydrostatic pressure

To probe the evolution of the ferroelectric polarization under pressure, polarization-voltage hysteresis loops of the CIPS single crystal were measured during the process of compression [Fig. 2(a)], from 0 to \approx 1.8 GPa. The remanent hysteresis technique was used to remove the contribution from non-remanent polarization components and leakage due to ionic conduction. The polarization value increases with the increment of pressure, consistent with the negative piezoelectricity of CIPS. In the early stage of pressurization, the polarization value rises rapidly, from $4.4 \,\mu\text{C/cm}^2$ at ambient pressure to about $6.0 \,\mu\text{C/cm}^2$ under 0.5 GPa [Fig. 2(b)]. Afterwards, the growth rate slows down. The polarization value slowly increases at a rate of about 0.7 μ C/cm² per GPa. This behavior is in good agreement with the trend of Cu¹ and Cu^O site fraction obtained from experimental XRD refinements [Fig. 1(f)], from which the average off-center displacement of the Cu ion can be observed [Fig. 1(g)]. The results indicate that the initial increase in the polarization is due to the sharp transfer of Cu^I ions to Cu^O site at the low-pressure range, whereas the slow increase in polarization at high pressure can be attributed to the volumetric effect. In this case, the relative displacement of Cu from the center of sulfur octahedron remains almost fixed (clamped-ion), but the volume of the lattice continues to shrink, leading to enhanced polarization (dipole moment per volume). Real and imaginary parts of relative dielectric permittivity are also measured as a function of pressure and plotted in temperature and frequency domains [Fig. S5(a)]. Figure 2(c) shows the temperature-dependent real part of the permittivity measured at a frequency of 950 kHz under a pressure of 0, 0.045, 0.09, 0.128 GPa, respectively. Within our measurement range, the Curie temperature (T_C) of CIPS rises linearly at a rate of about 353.6 K/GPa [Fig. 2(d)]. This pressure coefficient is much larger than previously reported,^{18,1} potentially due to the inaccuracy in the pressure calibration and highly hysteretic behavior in this low-pressure region. Since CIPS exhibits finite ionic conduction at room temperature, through the dielectric data, we can obtain the AC conductivity $\sigma = 2\pi f \varepsilon_0 \varepsilon_r''$ [Fig. S5(b)], where *f* is the measurement frequency, ε_r'' is the imaginary part of the relative permittivity of CIPS, and ε_0 is the vacuum permittivity. The DC conductivity σ_{dc} can be extracted by fitting the low-frequency part of the curve according to Jonscher's power law,²

$$\sigma = \sigma_{dc} + A(2\pi f)^s, \tag{1}$$

from which A is the pre-exponential constant and s is the power law exponent with 0 < s < 1. Finally, its activation energy E_A of the DC ionic conduction is given based on the Arrhenius equation as follows:²⁷

$$\sigma T = \sigma_0 \cdot e^{\frac{-La}{k_B T}},\tag{2}$$



FIG. 2. Pressure-dependent electrical measurements. (a) Remanent polarization hysteresis loops of CIPS crystal measured at different pressures. The measurements were carried out at room temperature with a triangular voltage waveform of 12.5 Hz. (b) Measured remanent polarization as a function of applied hydrostatic pressure. (c) The dielectric anomalies of real permittivity near the ferroelectric-paraelectric phase transition boundary measured under 0, 45, 90, 128 MPa with a frequency of 950 kHz. (d) The Curie temperature as a function of applied hydrostatic pressure. (e) The Arrhenius plot of the DC conductivity with extracted activation energies under 0, 45, 90, 128 MPa (below T_C). (f) Thermal parameters of Cu atoms under different pressures derived from the structural refinements.

and an Arrhenius plot demonstrates different E_A under different pressures [Fig. 2(e)]. The activation energy of CIPS increases with higher pressure below the Curie temperature. This means that compression inhibits ionic conductivity, which is also supported by the reduction of the leakage signal in the low-frequency polarization hysteresis loops under pressure [Fig. S5(c)]. The pressure-suppressed ionic conduction can be understood by an increased energy barrier for thermal hopping of the Cu ions due to a more compact crystal lattice, as also evidenced by the decrease in the thermal parameters U_{iso} (Cu) derived from the structural refinements [Fig. 2(f)]. The increased sharpness of the polarization hysteresis loop under pressure can also be explained by the suppression of the Cu hopping. The final rise of the U_{iso} (Cu) can be attributed to the proximity to critical pressure for the structural phase transition.

Vibrational spectrum of CIPS under high pressure

Raman spectroscopy was employed to probe kinetic changes of phonon modes of single crystal CIPS on increasing pressure. The primitive cell of CIPS contains two formula units with 20 atoms, which results in a total of 60 normal vibrational modes (including three acoustic modes). Based on the group-theoretical analysis, the irreducible representations for all optical vibrational modes at the Brillouin zone center (Γ) are²⁸

$$\Gamma_{\text{optic}} = 28\text{A}' + 29\text{A}'',\tag{3}$$

among which the non-degenerate A' (symmetric) and A" (antisymmetric) are both Raman and infrared active. We have calculated all the phonon modes and their corresponding scattering intensity based on calculated Raman tensors at 0 K, as summarized in Table S1. The simulated Raman spectrum (light polarization along the [110] direction) generally matches the experimental result (light polarization within ab plane) except for some red-shifts of the high-frequency modes [Fig. S6 (a)]. Notably, the low-frequency peak not predicted by the theory can be attributed to the out-of-plane component of atomic vibrations due to random light scattering at the crystal edges.²⁴ By adding some outof-plane electric field component, the low-frequency peak can be nicely reproduced by the simulated spectrum [Fig. S6(b)]. The majority of the Raman active peaks appear to blueshift on increasing pressure due to an overall hardening of the elastic constant of the chemical bonds induced by lattice compression [Fig. 3(a)]. Generally, the peak shift behavior can be classified into two categories: one group (73, 103, 164, 266, 375 cm^{-1} bands) shows a sudden rise after 0.2 GPa, whereas the other group (216, 239, 319 cm⁻¹ bands) softens first and hardens after 0.5 GPa [Figs. S7(c)-S7(e)]. The kink at 0.2-0.5 GPa can be attributed to the abrupt occupancy transfer from Cu^I to Cu^O site. The hardening of the former group can be understood by the enhanced interlayer interaction by more Cu^O occupancy. Through a careful analysis of the atomic motions for different phonon modes (supplementary material, gif images), it is found that the modes in the latter group all involve either the bending or stretching motion of the Cu-S3 bonds. During the Cu^I to Cu^O transition, the force constant of the Cu-S₃ bonds is possibly reduced, thus leading to a softening behavior of related vibrational modes.

When the pressure is greater than \approx 4 GPa, the number of Raman peaks decreases significantly, signaling the occurrence of the phase transition and higher symmetry of the high-pressure phase. This finding is consistent with the XRD results. In decompression branch, the

phase transformation occurs at lower pressure, suggesting first-order phase transition with hysteresis [Figs. S7(a) and S7(b)]. It is worth noting that a softening trend of the low-frequency peak is found, in stark contrast to other peaks [Fig. 3(b)]. This peak can be further deconvoluted into two peaks by Gaussian multi-peak fitting [Fig. 3(c) and Fig. S8], and we found completely opposite peak shifts for these two peaks with regard to pressure, namely, one hardens, while the other softens [Fig. 3(d)].

Our DFT calculations reveal that the main contribution of Raman peak at low frequency comes from mode #7 [Fig. 3(f)], that is, the opposite vibration of adjacent layers in the z direction, including the z direction vibration of Cu atoms and a slight interlayer sliding. The pressure-dependent behavior of this vibration mode appears to directly correlate with the Cu position within the lattice [Fig. 3(g)]. Below 1 GPa, the Cu remains inside the sulfur octahedron (Cu¹ site). Meanwhile, mode #7 continues to soften in this pressure range. Above 1 GPa, the Cu ion suddenly jumps outside the octahedral cage and into the vdW gap (Cu^O site). Correspondingly, mode #7 starts to harden with increasing pressure. Based on these calculation results, we can conclude that the two sub-peaks around $30 \, \text{cm}^{-1}$ correspond to the contribution of Cu^I and Cu^O site occupancies, respectively. The opposite pressure-dependent behaviors for these two sites result in the crossover of Raman peak position shifts shown in Fig. 3(d). To further corroborate this conclusion, we compared the ratio of Cu^O/Cu^I site occupancy derived from XRD and the peak intensity ratio of the two sites' contribution based on Raman spectrum. The two results are in good agreement with each other [Fig. 3(e)].

DISCUSSION

We now reconcile the experimental (mostly at 300 K) and calculation results (at 0 K) by considering the temperature effect and discuss in detail the relationship between Cu displacement and interlayer sliding during the pressurization and phase transition.

At 0 K, Cu completely occupies the Cu¹ site at atmospheric pressure. Under compression, the vibrational mode of Cu in the z direction continues to soften, resulting in an isosymmetric phase transition of the material—Cu changes from occupying the Cu^I site to completely occupying the Cu^O site over 1.0 GPa. Because the Cu^O has strong interaction with the adjacent sulfur S4, the z-direction vibration mode (#7 mode) of Cu, like other vibration modes, now continues to harden under subsequent compression. At about 5.8 GPa, a first-order phase transition driven by the interlayer sliding takes place, converting the crystal symmetry from monoclinic (Cc) to trigonal (P31c). Accompanying the phase transition is the movement of Cu back to the off-center position inside the sulfur octahedron. However, different from the Cu^I site of the Cc phase, this off-center site is closer to the center of the octahedral cage. The crystal structure and Cu site of the high-pressure phase is analogous to CuInP₂Se₆, which somewhat hints the close structural connection between these homologous compounds.29

At room temperature under atmospheric pressure, Cu ions show a broadening of its distribution toward the metastable Cu^O site in the vdW gap due to thermally activated hopping motion. Upon compression, the free energy of the Cu^O site is reduced [Fig. 4(b)], driving more Cu to displace from the intra-layer Cu^I site toward the in-gap Cu^O site. The occupancy of these two sites, thus, depends delicately on the balance between their relative energy landscape and the thermally activated hopping effect. As a result, a complete occupancy of the Cu^O

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FIG. 3. Pressure-dependent vibrational spectra. (a) Raman spectra of CIPS at different pressures. (b) Zoomed-in plot of the low-frequency part of the Raman spectra. (c) Detailed deconvolution of the lowfrequency Raman peaks by Gaussian multi-peak fitting. The peak around 30 cm⁻¹ can be separated into two components due to the contribution from Cu^O and Cu^I sites. (d) The peak positions of the two sub-peaks as a function of the pressure. (e) A comparison between the Cu^O/Cu^I occupancy ratio and the Raman peak intensity ratio as a function of the pressure. (f) A cartoon depicts the #7 A' mode. The blue arrows indicate the motions of Cu ions, while the purple ones denote the overall motions of each layer. (g) DFT calculated z position of the Cu ion and corresponding peak position of the #7 mode as a function of the pressure.

site was not seen at room temperature up to the highest pressure before phase transition. The conclusion is also confirmed by ferroelectric polarization measurements, where the polarization is in good agreement with the value calculated from a mixed occupancy of Cu^I (low polarization, $\approx 3.3 \,\mu\text{C/cm}^2$) and Cu^O (high polarization, $\approx 12.2 \,\mu\text{C/cm}^2$) sites¹² [Fig. S5(d)]. In conjunction with the polarization increase is the drastic enhancement of T_C under compression, consistent with the negative piezoelectricity of CIPS. For an order-disorder-type ferroelectric like CIPS, the T_C increment can be interpreted as the increase in the potential barrier for thermal hopping of Cu ions between two centrosymmetric off-center sites. Such

explanation is also supported by the rise of activation energy for ionic conduction under increased pressure. The weak softening of the #7 zone-center mode can also be attributed to the competition behavior of the Cu^O and Cu^I sites with regard to pressure. Interestingly, #7 mode involves the out-of-plane displacive motion of the Cu ions in association with the shear motion of the entire layer (supplementary material, gif images), suggesting the interlock coupling between Cu displacement and interlayer sliding as evidenced by the structural refinement results. In fact, through a group-subgroup analysis, it can be found that the *P*31*c* phase is transformed into *Cc* phase by two irreducible representations, Γ_1^+ and Γ_3^- (supplementary material, gif image), which involve



FIG. 4. Phase diagram and energy landscape in CIPS. (a) Pressure-temperature (P–T) phase diagram of CIPS constructed jointly by experimental and theoretical results. The dash-line phase boundaries are qualitatively determined by connection and extrapolation among the data points. The red and blue arrows denote two different structural transition as detailed in (b). PE: paraelectric, FE: ferroelectric, T_C: Curie temperature, P_C: transition pressure. (b) Schematic energy landscapes of Cu displacive transition and interlayer sliding transition tuned by hydrostatic pressure. The sliding distance *I* is defined as the in-plane interlayer displacement between two In ions from nearest layers. Here, *I* is normalized to the in-plane lattice constant a_0 of CuInP₂S₆, such that $||a_0 = 0$ ($||a_0 = 1/3$) explicitly corresponds to the P31c (Cc) phase.

out-of-plane motion of Cu and interlayer shearing, respectively. At the phase transition point, relative interlayer sliding by $-1/3 \vec{a}$ along the *a*-axis happens. Such a structural transformation from monoclinic to trigonal phase seems to be a universal behavior of metal thiophosphates,^{30–34} in which high lattice symmetry and more compact atomic packing possibly reduce the total energy under compressive pressure.

In the case of CIPS, the interlayer sliding causes the loss of bonding between Cu and the adjacent S4 atom. Consequently, the Cu ion at the Cu^O site becomes unstable, driving it back into the sulfur octahedral cage. In contrast to the polar phase (P31c) predicted at 0 K, we observed a non-polar phase $(P\overline{3}1c)$ at room temperature due to the disordered distribution of Cu among two off-center sites and the central site, which is, again, a reminiscence of the paraelectric phase of CuInP₂Se₆.²⁹ This suggests there is a ferroelectric-paraelectric phase transition at finite temperature for the high-pressure P31c phase of CIPS. The Curie temperature of P31c phase was estimated through Landau φ 4 potential, while Ginzburg term was not considered based on previous results on Cc phase.³⁵ Through analyzing its Boltzmann factor ($A_0e^{-Ep/kT}$), we determine that the T_C of the *P*31*c* phase is about 51 K. Here, the calculated energy barrier Ep between the ferroelectric P31c and non-polar $P\bar{3}1c$ phase is ≈ 28.7 meV/f.u. To simplify, the constant A_0 is estimated by fitting bulk Cc phase (T_C \approx 225 K, E_p \approx 120 meV/f.u.). By summarizing the experimental and theoretical findings, a pressure-temperature phase diagram of CIPS can be constructed as shown in Fig. 4(a). The phase diagram highlights the effect of pressure on the energy landscapes of Cu displacive transition in low-pressure region and interlayer sliding transition in high-pressure region. As shown in Fig. 4(b), the red box represents the displacive transition of the Cu ions in low-pressure region, and its corresponding energy profile change is shown in the right panel of the figure. Our nudged elastic band (NEB) calculations reveal that under ambient pressure (0 GPa), Cu ions prefer to stay inside the sulfur layer. However, under small pressure conditions (e.g., 1 GPa), Cu ions tend to move toward the vdW gap, as schematically shown in red box of Fig. 4(b). On the other hand, the blue box represents the interlayer

We note that a related report³⁶ was recently published during the preparation of our manuscript. However, the report by Yao *et al.* focused mainly on the polarization enhancement in the early stage of the pressurization, which is part of our results reported herein. Moreover, different from their report, we observed a continuous increase in polarization after 1.5 GPa, which is consistent with our structural results and previous study by second harmonic generation.²⁴ This results in a totally different interpretation of the Cu displacement behavior after 1.5 GPa.

CONCLUSION

Through a comprehensive analysis of combined experimental and theoretical results, this work unravels the complicated yet intriguing structural transformation of vdW layered ferroelectric CuInP₂S₆ under hydrostatic pressure. In low-pressure region (<0.5 GPa), the compression renders the Cu^O site energetically more favorable, thus resulting in an abrupt increase in the Cu^O site occupancy and consequently the polarization enhancement. However, due to the small

sliding transition of the whole layer, revealing an asymmetrical doublewell potential in the sliding potential curve [the bottom panel of Fig. 4(b)]. Specifically, the enthalpy of the Cc phase is lower than that of the P31c phase ($H_{Cc} - H_{P31c} = -0.21 \text{ eV/f.u.}$) under ambient pressure (0 GPa). Remarkably, as pressure increases beyond the transition point, for instance, enthalpy difference $H_{Cc} - H_{P31c} = 0.05 \text{ eV/f.u.}$ at 6 GPa, the P31c phase becomes more stable than the Cc phase. Furthermore, pressure has the effect of reducing the sliding barrier between the P31c and Cc phases, promoting the occurrence of the sliding transition. Importantly, all theoretical results are in excellent agreement with experimental findings. Additionally, our DFT calculation results indicate that the sign of the longitudinal piezoelectric coefficient (d_{33}) is directly associated with the Cu position in the lattice. The calculated d_{33} for the Cc phase with total Cu^I or Cu^O occupancy is -7.01pC/N and 1.6 pC/N, respectively, consistent with previous report.11 The value changes to 7.08 pC/N for the P31c phase. This finding signifies the importance of stacking order on the electromechanical behavior of the crystal lattice.

energy difference between these two sites, a full occupancy of Cu^O at room temperature is not realized by further pressurization due to thermally assisted hopping, prior to the structural phase transition driven by interlayer sliding. A soft mode with combined Cu translation and interlayer shear motions is identified to closely link with this slidingmediated phase transition. The pressure-induced structural evolution of CIPS pivots on the displacive instability of Cu ion, which serves as a bridging switch to modulate the interlayer interaction in response to external pressure. The interlayer sliding induced changes of crystal symmetry in conjunction with the ferroelectric polarization highlights again the critical role of interlayer stacking pattern in tailoring the long-range order parameters in vdW layer crystals.

METHODS

Materials synthesis

 $CuInP_2S_6$ single crystals were synthesized by chemical vapor transport (CVT) method as reported previously,^{10,13} with iodine as the transport agent. The composition stoichiometry and phase purity were checked by energy dispersive x-ray spectroscopy (EDS) and Raman spectroscopy. High purity and regular shape crystals are selected for subsequent measurements.

Single-crystal x-ray diffraction under high pressure

A pale yellow, platelet single crystal of CuInP_2S_6 characterized under ambient conditions was loaded into a Merrill–Bassett diamond anvil cell (DAC),³⁷ equipped with Boehlar–Almax diamond anvils, tungsten carbide backing seats³⁸ (opening angle = 40°) and a tungsten gasket. The sample chamber was filled with a 4:1 volumetric mixture of methanol:ethanol to serve as a pressure-transmitting medium, and the pressure within the sample chamber was measured using the ruby fluorescence method.³⁹

The crystal structure of CuInP₂S₆ was characterized under ambient conditions of temperature and pressure using a Rigaku Synergy-S diffractometer with mirror-chromated Mo K α radiation ($\lambda = 0.7107$ Å). Diffraction data were integrated and reduced in CrysAlis Pro⁴⁰ and an absorption correction was applied using SADABS.⁴¹ The crystal structure was solved using ShelXT⁴² and refined against $|F^2|$ using ShelXL⁴³ in Olex2 (Ref. 44) as a twocomponent inversion twin. All geometric and anisotropic displacement parameters were refined freely. The detailed procedures of data collection and structural refinement are described in supplementary material, Text 2. The refined crystallographic data are included in Tables S2 and S3.

Electrical measurements under high pressure

CuInP₂S₆ crystals with thicknesses of 10–20 μ m were selected to fabricate the parallel-plate ferroelectric capacitors with Au electrodes. The sample was then loaded into a hydrostatic pressure cell (HPC-33, Quantum Design) for high-pressure electrical tests. The sample was connected in series to the external circuit through the sample board and sample feedthrough of the instrument. A hydraulic press set (Model FHP-5P-37S/40S) was used to apply pressure, and the pressurizing medium is daphne 7373 oil. The upper pressure and temperature limit of the instrument is 3.0 GPa and 400 K, respectively.

A commercial ferroelectric tester (Radiant Technologies, Multiferroic II) was used to measure the standard polarization hysteresis loop and remanent hysteresis loop. Dielectric data upon different temperature under each pressure were collected by a commercial LCR meter (Agilent, 4284A) in conjunction with a temperature chamber (Model 9023, Delta Design).

Raman spectroscopy under high pressure

Raman spectroscopies were conducted on a confocal Raman microscope (WITec Alpha 300RAS) in backscattering geometry with an excitation laser wavelength of 633 nm (He–Ne laser) and an objective lens (×50) with a long working distance and two notch filters. Laser power was kept as high as 8 mW with the laser spot size of 1 μ m to obtain sufficient Raman signal. The high-pressure cell used in this experiment was a symmetric Mao-type DAC with two diamonds with 400 μ m culet size. The sample and pressure-transmitting medium (silicone oil) with a small ruby piece (≈10 μ m, for pressure calibration) were loaded into a hole of 150 μ m diameter drilled in a gasket (T301, 250 μ m thick) that had been pre-indented.

Computational method

The density-functional theory (DFT) calculations were performed using the plane wave basis as implemented in Vienna *ab initio* Simulation Package (VASP).⁴⁵ The generalized gradient approximation in the Perdew–Burke–Ernzerhof (GGA-PBE) formulation is adopted with a cutoff energy of 500 eV.⁴⁶ Brillouin zones are sampled using $6 \times 6 \times 3$ grid *k*-points in the Monkhorst–Pack scheme. The phonon frequencies are determined using density functional perturbation theory (DFPT), which is implemented in the VASP code. The Raman tensor is calculated through macroscopic dielectric tensor.⁴⁷ The evaluation of polarization is conducted using the Berry-phase method,⁴⁸ wherein both the ionic and electronic contributions are taken into account. The nudged elastic band (NEB) method is employed to generate and calculate a series of transition states between the ferroelectric and paraelectric states.⁴⁹

SUPPLEMENTARY MATERIAL

See the supplementary material for more information on the analyses of single-crystal XRD, dielectric and polarization measurements, Raman spectroscopy results, and DFT calculations.

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AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

Author Contributions

Zhou Zhou, Jun-Jie Zhang, and Gemma F. Turner contributed equally to this work.

Zhou Zhou: Data curation (lead); Investigation (equal); Methodology (equal); Validation (equal); Writing - original draft (equal); Writing review & editing (equal). Jun-Jie Zhang: Data curation (equal); Investigation (equal); Methodology (equal); Writing - original draft (equal). Gemma F. Turner: Data curation (equal); Formal analysis (equal); Investigation (equal); Methodology (equal); Writing - review & editing (equal). Stephen A. Moggach: Formal analysis (equal); Funding acquisition (equal); Supervision (equal); Validation (equal); Writing - original draft (equal); Writing - review & editing (equal). Yulia Lekina: Data curation (supporting); Formal analysis (supporting). Samuel Morris: Formal analysis (supporting); Methodology (supporting); Resources (supporting). Shun Wang: Data curation (supporting); Methodology (supporting). Yiqi Hu: Data curation (supporting). Qiankun Li: Data curation (supporting). Jinshuo Xue: Methodology (supporting). Zhijian Feng: Data curation (supporting). Qingyu Yan: Data curation (supporting). Yuyan Weng: Project administration (supporting); Supervision (supporting). Bin Xu: Methodology (supporting); Validation (supporting). Yong Fang: Methodology (supporting); Resources (supporting). Ze Xiang Shen: Project administration (supporting); Resources (supporting); Supervision (supporting). Liang Fang: Funding acquisition (equal); Project administration (equal); Supervision (equal); Writing - review & editing (equal). Shuai Dong: Conceptualization (equal); Formal analysis (equal); Funding acquisition (equal); Project administration (equal); Supervision (equal); Writing - review & editing (equal). Lu You: Conceptualization (equal); Funding acquisition (equal); Project administration (equal); Supervision (equal); Writing - original draft (equal); Writing – review & editing (equal).

DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding authors upon reasonable request.

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